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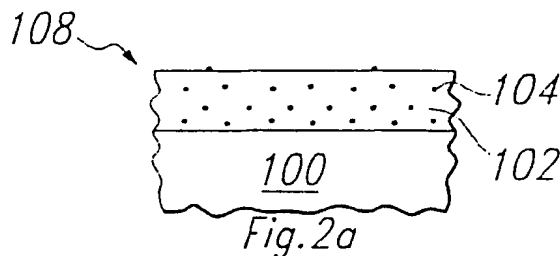
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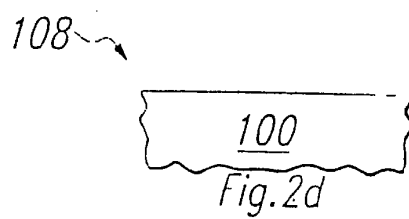
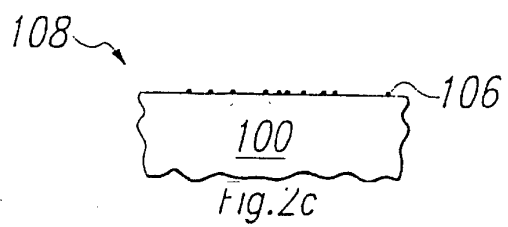
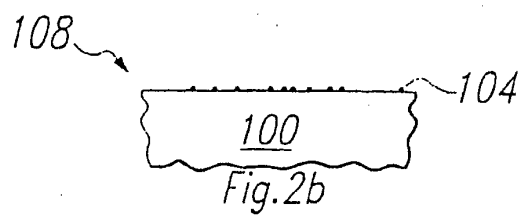
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(54) Improvements in or relating to the cleaning of semiconductor devices

(57) An embodiment of the instant invention is a method of removing inorganic contamination (contamination 104 of FIGURES 2a-2b) from a layer (layer 102) overlying a substrate (substrate 100), the method comprising the steps of: removing the layer overlying the substrate with at least one removal agent; reacting the inorganic contamination with at least one conversion

agent, thereby converting the inorganic contamination; removing the converted inorganic contamination by subjecting it to at least one solvent agent, the solvent agent included in a first supercritical fluid; and wherein the converted inorganic contamination is more highly soluble in the solvent agent than the inorganic contamination.





Description

FIELD OF THE INVENTION

The instant invention pertains to semiconductor device fabrication and processing and more specifically to a method of removing inorganic contamination in a supercritical fluid media.

BACKGROUND OF THE INVENTION

In the manufacture of, for example, integrated circuits and liquid crystal displays, contamination of the substrate and subsequent semiconductor layers causes many problems and should be reduced as much as possible. Examples of such contamination are residual particulates, organics, and metals. In addition, contaminants may be situated on the surface of the semiconductor layer, or they may be situated between the semiconductor layer and another layer (e.g. an oxide layer). Typically, wet-processing is utilized in the fabrication of semiconductor devices. A wet cleaning process may consist of a sequence of particle removal and metal removal steps, with a rinsing step performed between them, and concluding with a drying step. Drying is typically accomplished either by spinning the wafer so that the liquid on the wafer will be spun off or by raising the wafers from a cloud of hot isopropyl alcohol vapor which condenses on the surface of the wafer and displaces the water.

This type of wet cleaning process suffers from a particularly serious drawback, specifically, most metal removal processes of this type (typically consisting of strongly acidic mixtures) add particles to the surface of the wafer, and particle removal processes (typically comprised of basic/oxidant mixtures) add metals to the surface of the wafer. In addition, most wet cleaning processes suffer from the problems of: expense of obtaining electronic-grade wet chemicals; expense of disposing of the caustic chemicals which are used in wet cleaning processes; liquid surface tension limiting or preventing access of wet chemical cleaning to high aspect ratio features, like trenches; and lack of harmonization with all dry processing (which is utilized more frequently in semiconductor processing). Hence, a metal removal step followed by a particle removal step will result in residual metals, while the reverse process results in fewer particles but can cause contamination with metals by the final rinsing.

Supercritical fluids (e.g., supercritical carbon dioxide) have received much attention, lately. This is especially true in the areas involving decaffeination of coffee and linen/linen clothing dry cleaning. In addition, supercritical CO₂ can be used in the removal of organic contaminants from semiconductor wafers. See International Journal of Environmentally Conscious Design & Manufacturing, Vol. 2, p. 83 (1993) (stating "supercritical carbon dioxide is best applied to the removal of organic

compounds with mid-to-low volatilities"). However, supercritical CO₂ is generally regarded as ineffective in cleaning inorganic contaminants (e.g., metals) from semiconductor wafers.

Relating to another field, a group of researchers have discovered a way of removing metals from vegetation by subjecting the vegetation to supercritical CO₂ and neutralizing the metals with chelating agents. See Elizabeth K. Wilson, Toxic Metals Extracted with Supercritical Carbon Dioxide, C&EN 27 (April 15, 1996); and U.S. Patent No. 5,356,538. However, this work states that "by itself, non-polar supercritical CO₂ is almost useless for solvating positively charged heavy-metal ions. However, researchers have discovered that metals can be solvated if they are first neutralized by chelating agents and, furthermore, that the solvency increases dramatically when the chelating agents are fluorinated." Id. at 27. However, there are several problems with this approach. First, there is a difficulty removing uncharged metals. Second, unfluorinated chelating agents are expensive. Third, mass scale synthesis of fluorinated chelating agents is expensive. Fourth, fluorinated and unfluorinated chelating agents are highly toxic and expensive to purify and dispose of. Fifth, there is a limited range of metals that can be readily solvated by the fluorinated chelating agents. Sixth, diffusion of unchelated metals into an underlying semiconductor substrate would be disastrous using the methodology of this publication.

It is, therefore, an object of the instant invention to provide a method of removing metal contaminants from between a native oxide layer and a semiconductor wafers. It is another object of the instant invention to provide a method of removing inorganic contaminants from between a native oxide layer and semiconductor wafers.

SUMMARY OF THE INVENTION

In summary, one embodiment of the instant invention is a method which overcomes the problems involved with chemical modification of the ionic and neutral, light and heavy inorganic (metal) species and which renders these ionic and neutral, light and heavy inorganic (metal) species soluble upon exposure to conventional, inexpensive, high purity, non-toxic solvents. The method of the instant invention includes the steps of: removing the native oxide in which overlies the inorganic contamination (and/or that is situated between the inorganic contamination and the substrate and/or that surrounds the inorganic contamination), thereby exposing the inorganic contamination (that is contained within the native oxide or that underlies the native oxide); chemically altering the inorganic contamination; exposing the chemically-altered inorganic contaminant to a conventional solvent which is included in a supercritical fluid (preferably supercritical CO₂); and removing the conventionally-solvated, chemically-altered inorganic contaminant in a supercritical fluid (SCF). The chemical al-

tering of the inorganic contaminant may occur prior to or during the exposure to the SCF. Key aspects of this invention are: inorganic contamination within a native oxide is exposed for subsequent modification and removal; the inorganic contaminant is not soluble in the supercritical CO₂ fluid without prior chemical alteration; and the chemically-altered inorganic contaminant may be removed by the solvent simultaneously with the chemical alteration step.

An embodiment of the instant invention is a method of removing inorganic contamination from a layer overlying a substrate, the method comprising the steps of: removing the layer overlying the substrate with at least one removal agent; reacting the inorganic contamination with at least one conversion agent, thereby converting the inorganic contamination; removing the converted inorganic contamination by subjecting it to at least one solvent agent, the solvent agent included in a first supercritical fluid; and wherein the converted inorganic contamination is more highly soluble in the solvent agent than the inorganic contamination. Preferably, the conversion agent is selected from the group consisting of: an acid agent, a basic agent, a chelating agent, a ligand agent, a halogen-containing agent, and any combination thereof. More preferably, the conversion agent is comprised of HF and is included in supercritical CO₂. Preferably, the solvent agent is selected from the group consisting of: a polar gas, a non-polar gas, a polar supercritical fluid, a non-polar supercritical fluid, a polar species, a non-polar species, a surfactant, a detergent, an amphoteric material, or a chelating agent, and the solvent agent is included in supercritical CO₂. The overlying layer may be comprised of a native oxide. The steps of removing the layer overlying the substrate with a removal agent and reacting the inorganic contamination with a conversion agent may be performed simultaneously. Alternatively, the steps of removing the layer overlying the substrate with a removal agent, reacting the inorganic contamination with a conversion agent, and removing the converted inorganic contamination are performed simultaneously; or the step of removing the layer overlying the substrate with a removal agent is performed followed by the steps of reacting the inorganic contamination with a conversion agent and removing the converted inorganic contamination, which are performed simultaneously. Preferably, the removal agent is comprised of HF, and it is included in a second supercritical fluid. The second supercritical fluid is, preferably, comprised of supercritical CO₂.

Another embodiment of the instant invention is a method of removing inorganic contamination from a layer overlying a substrate, the method comprising the steps of: removing the layer overlying the substrate with at least one removal agent, the removal agent included in a first supercritical fluid; reacting the inorganic contamination with at least one conversion agent thereby converting the inorganic contamination, the conversion agent included in a second supercritical fluid; removing

the converted inorganic contamination by subjecting it to at least one solvent agent, the solvent agent included in a third supercritical fluid; and wherein the converted inorganic contamination is more highly soluble in the solvent agent than the inorganic contamination. The steps of removing the layer overlying the substrate with a removal agent and reacting the inorganic contamination with a conversion agent may be performed simultaneously. However, the steps of removing the layer overlying the substrate with a removal agent, reacting the inorganic contamination with a conversion agent, and removing the converted inorganic contamination may all be performed simultaneously. Alternatively, the step of removing the layer overlying the substrate with a removal agent is performed followed by the steps of reacting the inorganic contamination with a conversion agent and removing the converted inorganic contamination, which are performed simultaneously.

BRIEF DESCRIPTION OF THE DRAWINGS

The present invention will now be further described, by way of example, with reference to the accompanying drawings in which:

FIGURE 1 is a schematic drawing of a sample cleaning system in accordance with an embodiment of the instant invention; and

FIGURES 2a-2d are cross sectional views illustrating an embodiment of the instant invention.

DETAILED DESCRIPTION OF THE INSTANT INVENTION

FIGURE 1 illustrates a process system which can be utilized to implement the method of the instant invention. The sample to be cleaned (the semiconductor wafer with inorganic contamination present) is held in container 16. A supercritical fluid (preferably a CO₂ gas) is supplied from a gas reservoir 28 which is connected by conduit 30, which includes valve 32, to a pressurization unit 34 that increases the pressure on the gas to greater than approximately 70 to 75 atmospheres at a temperature greater than approximately 32 C to form a supercritical fluid. The supercritical fluid (SCF) travels through valve 36 and conduit 38 to a reservoir 11 (as long as valves 1 and 3, 5 and 9 are open and valve 2 is closed) that holds a solid, liquid, or gas removal agent(s). Possible removal agents are listed below. The passing of the SCF through the removal agent acts to incorporate the removal agent into the SCF. The SCF incorporated with the removal agent leaves reservoir 11 and enters container 16. The SCF mixture and inorganic contamination are introduced, resulting in the removal of the top layer containing the inorganic contamination, thereby exposing the inorganic contamination (and possibly simultaneously modifying the inorganic contamination).

Subsequent to or simultaneous with removal of the

104 is comprised of sodium, and that this contamination is interdispersed throughout native oxide 102 (as depicted in FIGURE 2a). In order to remove native oxide 102, Hydrofluoric acid (HF) is introduced to device 108. The HF can either be incorporated in a SCF or not. The result of this step is shown in FIGURE 2b. Next, device 108 is exposed to HF (this may be done simultaneously with the previously mentioned step). This step results in the conversion/alteration of the sodium contamination. The result is NaF (referred to as contamination 106 in FIGURE 2c). Next, device 108 is subjected to a solvent (water incorporated in supercritical CO₂) so as to remove the more soluble contamination 106. The result is shown in FIGURE 2d, whereby both the native oxide and the contamination are removed.

Although specific embodiments of the present invention are herein described, they are not to be construed as limiting the scope of the invention. Many embodiments of the present invention will become apparent to those skilled in the art in light of methodology of the specification.

Claims

1. A method of removing inorganic contamination from a layer overlying a substrate, said method comprising the steps of:
 - removing said layer overlying said substrate with at least one removal agent;
 - reacting said inorganic contamination with at least one conversion agent, thereby converting said inorganic contamination;
 - removing said converted inorganic contamination by subjecting it to at least one solvent agent, said solvent agent included in a first supercritical fluid;
 - and wherein said converted inorganic contamination is more highly soluble in said solvent agent than said inorganic contamination.
2. The method of Claim 1, further comprising selecting said conversion agent from the group comprising: an acid agent, a basic agent, a chelating agent, a ligand agent, a halogen-containing agent, and combinations thereof.
3. The method of Claim 1 or Claim 2, wherein said reacting step comprises reacting said inorganic contamination with a conversion agent comprised of HF.
4. The method of any preceding claim, wherein said reacting step comprises reacting said inorganic contamination with a conversion agent comprised in supercritical CO₂.
5. The method of any preceding claim, further comprising selecting said solvent agent from the group comprising: a polar gas, a non-polar gas, a polar supercritical fluid, a non-polar supercritical fluid, a polar species, a non-polar species, a surfactants, a detergent, a amphoteric material, or a chelating agent.
6. The method of any of Claims 1 to 4, wherein said removing step comprises removing said inorganic contamination with a solvent agent comprised in supercritical CO₂.
7. The method of any preceding claim, further comprising performing said steps of removing said layer overlying said substrate with a removal agent and reacting said inorganic contamination with a conversion agent simultaneously.
8. The method of any of Claims 1 to 6, further comprising performing said steps of removing said layer overlying said substrate with a removal agent, reacting said inorganic contamination with a conversion agent, and removing said converted inorganic contamination simultaneously.
9. The method of any of Claims 1 to 6, further comprising performing said step of removing said layer overlying said substrate with a removal agent followed by performing the steps of reacting said inorganic contamination with a conversion agent and removing said converted inorganic contamination simultaneously.
10. The method of any preceding claim, wherein said step of removing said layer overlying said substrate is performed with removal agent comprised of HF.
11. The method of Claim 10, further comprising including said removal agent in a second supercritical fluid.
12. The method of Claim 11, wherein said including step comprises including said removal agent in second supercritical fluid comprised of supercritical CO₂.
13. A method of removing inorganic contamination from a layer overlying a substrate, said method comprising the steps of:
 - removing said layer overlying said substrate with at least one removal agent, said removal agent included in a first supercritical fluid;
 - reacting said inorganic contamination with at least one conversion agent thereby converting said inorganic contamination, said conversion agent being included in a second supercritical

top layer containing the inorganic contamination by the removal agent (thereby exposing the inorganic contamination) and subsequent to or simultaneously with removing the modified inorganic contamination, the SCF travels through valve 36 and conduit 38 to reservoir 12 which holds a solid, liquid, or gaseous modification agent(s). Possible modification agents are listed below. This is accomplished by closing valves 1, 3, and 5 and opening valves 2, 4, 6 and 9. The passing of the SCF through the modification agent acts to incorporate the modification agent into the SCF. The SCF incorporated with the modification agent leaves reservoir 14 and enters chamber 16. The SCF mixture and the exposed inorganic contamination are introduced, thereby resulting in the modification of the inorganic contamination, which is situated on the surface of the sample (preferably a semiconductor wafer).

Subsequent to or simultaneous with the removal of the top layer (thereby exposing the inorganic contamination) and subsequent to or simultaneously with the modification of the inorganic contamination on the semiconductor sample by the modification agent, the SCF travels through valve 36 and conduit 38 to reservoir 14 which holds a solid, liquid, or gaseous solvent agent(s). Possible solvent agents are listed below. This is accomplished by closing valves 1, 3, 4, 6, and 9 and opening valves 2, 5, and 8. The passing of the SCF through the solvent agent acts to incorporate the solvent agent into the SCF. The SCF incorporated with the solvent agent leaves reservoir 14 and enters chamber 16. The SCF mixture and the exposed and modified inorganic contamination are introduced, thereby resulting in the removal of the exposed and modified inorganic contamination from the surface of the sample (preferably a semiconductor wafer).

The modified inorganic contaminant and the CO₂ are removed and passed through depressurization valve 18 such that the inorganic contaminant precipitates in container 20. The CO₂ gas is then recycled by pump 24 through line 26 to reservoir 28. The inorganic contaminants can be removed via line 22.

One embodiment of the instant invention is a method of removing inorganic (preferably metallic) contamination from within a native oxide layer or from between a underlying semiconductor layer and a layer of native oxide. Preferably, this method is comprised of the following steps. First, removing the native oxide (which is of the order of 30 Angstroms thick) by subjecting it to a removal agent. Second, reacting the inorganic contamination with a conversion agent so as to convert the inorganic contamination (preferably in a more soluble form). The removal agent and the conversion agent may be comprised of the same elements. Third, removing the converted inorganic contamination with a solvent agent. The removal agent, the modification agent, and the solvent agent may be comprised of the same element(s) and may be administered simultaneously or sequentially.

The removal agent may be comprised of hydrofluoric acid. In addition, it may be introduced either by vapor exposure, plasma exposure, or by exposing the semiconductor wafer to a supercritical fluid (preferably CO₂) which contains HF. The conversion agent may be comprised of HF or it may include any other halogen-containing agent (preferably chlorine). The conversion agent may be introduced by vapor exposure to the wafer, plasma exposure to the wafer, or by exposing the wafer to a supercritical fluid (preferably CO₂) which includes the conversion agent. Preferably, the conversion agent is comprised of an acid (preferably HCN, KCN, HF, HCl, or HI or KI), a base (preferably NH₄OH, KOH, or NF₃), a chelating agent (preferably dibetaketone), or a halogen-bearing agent (preferably HF, F₂, Cl₂, Br₂, or I₂) or a polar agent (preferably CO, NH₃, NO, COS, NH₄OH, water, or H₂O₂). Preferably, the solvent agent may be comprised of a polar gas (preferably CO, COS, NO, NH₃, or NF₃), non-polar gases (preferably N₂, H₂, O₂, or F₂), polar SCF (preferably NO₂), non-polar SCF (preferably CO₂), a polar species (preferably water, ethanol, methanol, acetone, or glycol), a non-polar species (preferably tetrahydrofuran, or dimethylformamide), surfactants, detergents, or amphoteric materials (preferably sodium dodecyl sulfate, quaternary ammonium salts, or cationic, anionic, nonionic or zwitterionic surfactants), or a chelating agent (preferably beta-diketone, fluorinated or unfluorinated crown ether), which is preferably included in a supercritical fluid (preferably CO₂).

Referring to FIGURES 2a-2d, the method of this embodiment is capable of removing inorganic contamination 104 that is situated on top of a native oxide 102, within the native oxide 102, or between native oxide 102 and underlying layer 100. This method can be utilized to remove the native oxide layer 102 (see FIGURES 2a and 2b), convert the inorganic contamination so that it is more soluble in a subsequent removal step (see FIGURES 2b and 2c where inorganic contamination 104 is modified to a more soluble inorganic contamination 106), and removing the converted inorganic contamination (see FIGURE 2d). The conversion agent may be comprised of any agent that would make the inorganic contamination more soluble (or more easily removed) in a subsequent removal step. Therefore, the conversion agent may include a chelating agent. In addition, the removal of the native oxide and the conversion of the inorganic contamination may occur in one step.

The removal agent may or may not be included in a SCF, and the conversion agent may or may not be included in a SCF. In addition, the removal agent, the conversion agent, and the solvent agent may all be introduced at one time; or the removal agent and the conversion agent may be introduced together followed by the solvent agent; or the removal agent may be introduced followed by the combination of the conversion agent and the solvent agent.

For example, assume that inorganic contamination

fluid;
removing said converted inorganic contamination by subjecting it to at least one solvent agent, said solvent agent included in a third supercritical fluid;
and wherein said converted inorganic contamination is more highly soluble in said solvent agent than said inorganic contamination.

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14. The method of Claim 13, further comprising performing said steps of removing said layer overlying said substrate with a removal agent and reacting said inorganic contamination with a conversion agent simultaneously.

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15. The method of Claim 13, further comprising performing said steps of removing said layer overlying said substrate with a removal agent, reacting said inorganic contamination with a conversion agent, and removing said converted inorganic contamination simultaneously.

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16. The method of Claim 13, further comprising performing said step of removing said layer overlying said substrate with a removal agent and followed by performing the steps of reacting said inorganic contamination with a conversion agent and removing said converted inorganic contamination simultaneously.

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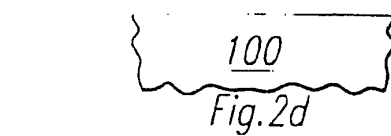
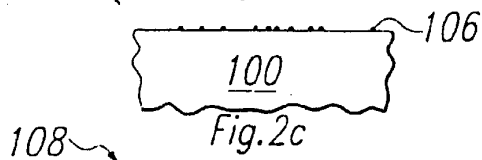
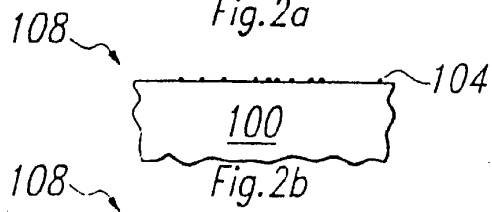
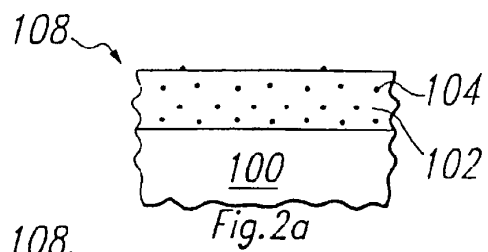
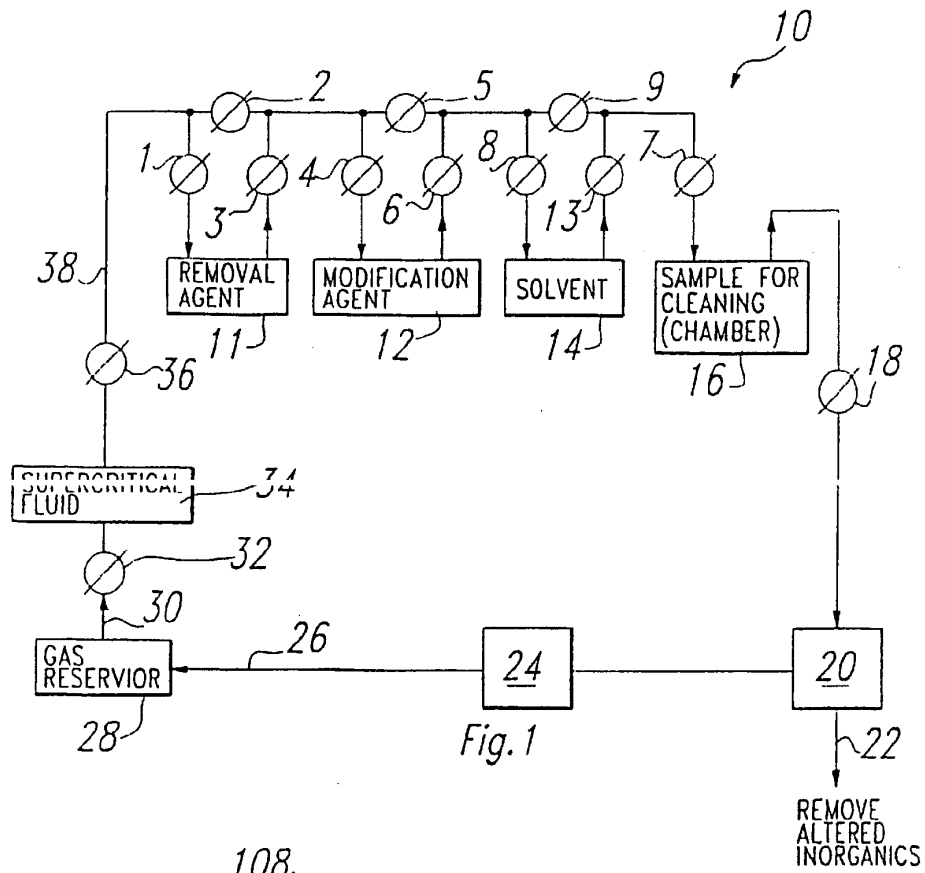
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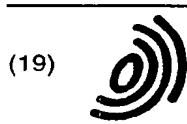
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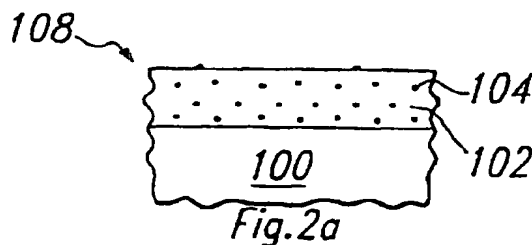
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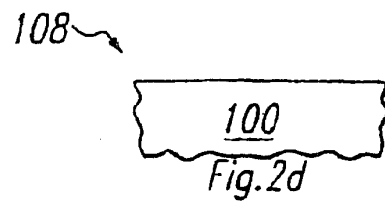
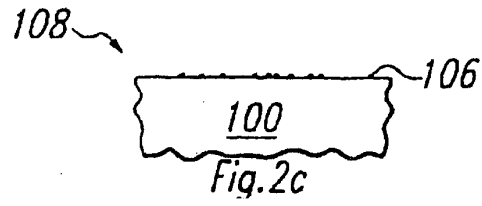
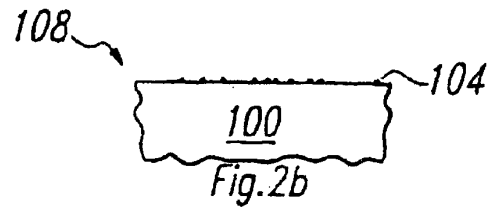
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(54) Improvements in or relating to the cleaning of semiconductor devices

(57) An embodiment of the instant invention is a method of removing inorganic contamination (contamination 104 of FIGURES 2a-2b) from a layer (layer 102) overlying a substrate (substrate 100), the method comprising the steps of: removing the layer overlying the substrate with at least one removal agent; reacting the inorganic contamination with at least one conversion

agent, thereby converting the inorganic contamination; removing the converted inorganic contamination by subjecting it to at least one solvent agent, the solvent agent included in a first supercritical fluid; and wherein the converted inorganic contamination is more highly soluble in the solvent agent than the inorganic contamination.







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EUROPEAN SEARCH REPORT

Application Number
EP 97 30 5822

DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.6)
X	US 5 105 556 A (HITACHI) * column 5, line 59 - line 64; example 7 *	1-8. 10-15	H01L21/306 B08B3/10
Y	---	9.16	
Y	PATENT ABSTRACTS OF JAPAN vol. 18, no. 545 (E-1617), 18 October 1994 & JP 06 196472 A (SOLTEC K. K.), 15 July 1994, * abstract *	9.16	
X	EP 0 391 035 A (HUGHES AIRCRAFT) * claims *	1-16	
X	PATENT ABSTRACTS OF JAPAN vol. 13, no. 507 (E-845), 14 November 1989 & JP 01 204427 A (HITACHI), 17 August 1989, * abstract *	1-16	
A	EP 0 536 752 A (AIR PRODUCTS AND CHEMICALS) * abstract *	2	TECHNICAL FIELDS SEARCHED (Int.Cl.6) H01L B08B
A	US 5 352 327 A (WITOWSKI ET AL) * abstract *	1-16	
The present search report has been drawn up for all claims.			
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<div> <div> <p>DATE WHEN THE LISTED DOCUMENTS</p> <ul style="list-style-type: none"> 1. were made available to the applicant 2. were made available to the public 3. were made available to the public 4. were made available to the public 5. were made available to the public 6. were made available to the public 7. were made available to the public 8. were made available to the public 9. were made available to the public 10. were made available to the public 11. were made available to the public 12. were made available to the public 13. were made available to the public 14. were made available to the public 15. were made available to the public 16. were made available to the public 17. were made available to the public 18. were made available to the public 19. were made available to the public 20. were made available to the public 21. were made available to the public 22. were made available to the public 23. were made available to the public 24. were made available to the public 25. were made available to the public 26. were made available to the public 27. were made available to the public 28. were made available to the public 29. were made available to the public 30. were made available to the public 31. were made available to the public 32. were made available to the public 33. were made available to the public 34. were made available to the public 35. were made available to the public 36. were made available to the public 37. were made available to the public 38. were made available to the public 39. were made available to the public 40. were made available to the public 41. were made available to the public 42. were made available to the public 43. were made available to the public 44. were made available to the public 45. were made available to the public 46. were made available to the public 47. were made available to the public 48. were made available to the public 49. were made available to the public 50. were made available to the public 51. were made available to the public 52. were made available to the public 53. were made available to the public 54. were made available to the public 55. were made available to the public 56. were made available to the public 57. were made available to the public 58. were made available to the public 59. were made available to the public 60. were made available to the public 61. were made available to the public 62. were made available to the public 63. were made available to the public 64. were made available to the public 65. were made available to the public 66. were made available to the public 67. were made available to the public 68. were made available to the public 69. were made available to the public 70. were made available to the public 71. were made available to the public 72. were made available to the public 73. were made available to the public 74. were made available to the public 75. were made available to the public 76. were made available to the public 77. were made available to the public 78. were made available to the public 79. were made available to the public 80. were made available to the public 81. were made available to the public 82. were made available to the public 83. were made available to the public 84. were made available to the public 85. were made available to the public 86. were made available to the public 87. were made available to the public 88. were made available to the public 89. were made available to the public 90. were made available to the public 91. were made available to the public 92. were made available to the public 93. were made available to the public 94. were made available to the public 95. were made available to the public 96. were made available to the public 97. were made available to the public 98. were made available to the public 99. were made available to the public 100. were made available to the public </div> </div>			

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